

Influence of Sulphite on the Drying Kinetics of Pumpkin Fruit Slices

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To cite this article:

Comfort Mkpentseen Bunde-Tsegba, Charles Chukwuma Ariahu, Bibiana Dooshima Igbabul, Joseph Oneh Abu. Influence of Sulphite on the Drying Kinetics of Pumpkin Fruit Slices. *World Journal of Food Science and Technology*. Vol. 3, No. 3, 2019, pp. 32-39.

doi: 10.11648/j.wjfst.20190303.11

Received: July 30, 2019; Accepted: October 26, 2019; Published: December 2, 2019

Abstract: Drying kinetics, of pumpkin fruit slices as influenced by sulphiting, was investigated. Pumpkin fruits, were sulphited at 0, 1.0, 1.5, 2.0 and 2.5% to obtain 4 mm thick (P0, 4; P1, 4; P1.5 4; P2, 4; P2.5, 4 - and 5 mm samples (P0, 5; P1, 5; P1.5, 5; P2, 5; P2.5, 5 respectively). Samples were dried at 60, 65, 70, 75 and 80°C at an air velocity of 1.53 m/s. Drying was carried out to constant moisture. Samples were screened using descriptive sensory evaluation to obtain samples P1, 4 and P2, 5 each dried at 75°C as the best samples for 4 mm and 5 mm respectively. Drying curves, moisture diffusivity, activation energy, drying time and rehydration capacity were determined. The drying curves obtained showed results for a short constant rate followed by a falling rate period. The effective moisture diffusivity varied from 6.235×10^{-11} to $12.808 \times 10^{-11} \text{m}^2/\text{s}$ for the 4 mm and 9.046×10^{-11} to $21.330 \times 10^{-11} \text{m}^2/\text{s}$ for the 5 mm samples. Activation energy obtained for P0, 4; P1, 4; P0, 5 and P2, 5 were 31.342, 32.292, 31.525 and 29.88 kJ/mol. respectively. Sulphiting reduced drying time at 1% level from 16.5 to 15.2 hours for the 4 mm sample and 17.8 to 16.9 hours at 2% level of sulphiting for the 5 mm. Sulphite treatment reduced rehydration capacity significantly at $p > 0.05$. This study implies that the parameter which governed the internal transfer of moisture was moisture diffusion.

Keywords: Drying, Sulphite, Diffusivity, Rehydration and Pumpkin

1. Introduction

Cucurbita pepo is known as pumpkin in English, *elegede* in Yoruba (Nigeria), *calbaza*, in Mexico and *furum* in Tiv (Nigeria). *Cucurbita pepo* is a herb grown for its fruits and edible seeds [1]. Pumpkin belongs to the family of cucurbitaceae which includes; cucumbers, melons, squash and gourds [2]. Cultivated pumpkins are believed to have originated from Central Africa, as far back as 550 BC and are now widely grown all over the world. Sulphiting of fruits and vegetables has been used in inhibiting enzymatic and non-enzymatic browning [3].

Fruits and seeds of pumpkin are mostly used for culinary and medicinal purposes [4]. In Nigeria, pumpkin (fruit) is commonly consumed fresh. The succulent fruits are boiled and taken alone or mashed with boiled yam to form potage or

the fruits are processed by peeling, removing the seeds, slicing and sun-drying. The dried slices are prepared into soup in combination with ground melon seeds (egusi) or beniseed as a delicacy. Pumpkin fruit has antioxidant capacities as a result of its carotenoids content. Carotenoids in terms of human health reduce cataracts and macular degeneration [5]. Carotenoids are the natural plant pigments which are responsible for the orange colour of pumpkins. It is considered to be a good source of carotenoids, and a very rich source of carotene [6]. The main nutrient is beta carotene, which generates vitamin A in the body. The major carotenoids in pumpkin fruits include; β -carotene, α -carotene, lutein, lycopene, cryptoxanthin and zeaxanthin [7]. Pumpkin fruit is also a valuable source of vitamins such as vitamin B₁ (thiamine), vitamin B₂ (riboflavin), vitamin B₃ (niacin), vitamin B₅ (pantothenic acid), vitamin B₆ (pyridoxine), vitamin B₉ (folic acid). However, the perishable

nature of this fruit limits its utilization.

Pumpkin fruits are very difficult to preserve. Freezing results to 'freeze burn', while refrigerating only keeps for a few days. The only method of preservation that is suitable is sun-drying. This is only common in the tropics and sub-tropics. In addition, sun-drying has a lot of limitations [8]. Dehydrated food products according to Serenet *et al.*, are hygroscopic, therefore, needs adequate drying [9]. Pumpkin fruit slices as a semi-finished fruit product needs to be crispy, in addition, have the potentials to reconstitute as close to the fresh form as possible when immersed in water

Studies by Gujral *et al.*, Ramallo and Mascherni revealed that drying is the most important process of preserving fruits and vegetables [10, 11]. Literature report indicates that a host of these drying processes employ convective hot air drying and about 85% of industrial dryers are convective hot air types [12, 13]. According to Vazquez-Vila, the removal of moisture prevents the growth and reproduction of microorganisms causing decay and inhibits many of the moisture-mediated deteriorative reactions. It brings about substantial reduction in weight and volume, thereby, minimizing packaging, storage, transportation and distribution cost. In addition, drying aids storability of the product under ambient conditions [14].

Sun drying has been used to dry fruits and vegetables in tropical and subtropical regions. Studies by Prajapati *et al.* revealed that the quality of fruits and vegetables dried by conventional sun drying technique has to be replaced with mechanical drying (dehydration) which is assumed to produce better and consistent quality fruits and vegetable products, taking less time and minimizing microbial spoilage and chemical deterioration, thereby, alleviating the problem of fruits and vegetable losses [15]. Nguyen and Price revealed that, use of hot air drying as an alternative drying method both decreases the drying time and improve the quality of the final product in addition to low cost [16]. Mechanical drying is therefore used for various foods such as banana. Several researchers used it to dry strawberry, steamed glutinous rice, oyster mushroom and coffee as well as combing with microwave to dry brown rice [17, 18, 19, 20, 21]. In dehydration process, hot air heats the samples to remove the moisture. In high moisture foods such as fruits and vegetables, the drying air removes water from the food surface by evaporation for the fact that the available water in the food sample under drying occur at a very high water activity. The controlling factor of the drying process at this point is the heat supplied by the drying air to the drying food sample. This stage of drying process is referred to as the constant rate period [16]. The removal of water from food products depends on not only drying temperature but also slice thickness, drying conditions as well as sample geometry [22, 23, 16]. When the water activity of the available water on the surface of a drying food sample drops, moisture diffuses from inside the food material to the surface as a result of partial pressure or concentration

gradient. The controlling factor of the drying processes (evaporation) then changes to the rate at which water travels through the thickness of the food to the surface. This marks the beginning of the falling rate period. Several works have been carried out on drying kinetics of different species of pumpkin fruits using variable slice thicknesses [24].

Effective Moisture Diffusivity (Deff) and Activation Energy (Ea)

In drying kinetics, moisture can be expressed in terms of moisture ratio which is the ratio of the removed water to the removable water in a food substance under drying. This is expressed as:

$$MR = \frac{M_t - M_e}{M_i - M_e} \quad (1)$$

Where M_i = initial moisture content, M_t = moisture at time t , M_e = equilibrium moisture content. Drying of fruits and vegetables is considered in the falling rate period [25]. Ficks' diffusion equation was used by Jittanit and Arumuganathan *et al.* to model the drying behaviour of food systems since drying in the falling rate period is controlled by internal moisture diffusion of the food material [26, 27]. Effective moisture diffusivity represents the conductive term of all moisture transfer mechanisms. Investigation by Roberts *et al.* revealed that, based on assumptions of initial uniform moisture content, negligible external resistance to mass transfer, negligible temperature gradients, negligible shrinkage during drying and constant diffusivity coefficient, the analytical solution of diffusion equation is given for an infinite slab as:

$$\leftarrow MR = \frac{M_t - M_e}{M_o - M_e} = \frac{8}{\pi^2} \exp\left(-\pi^2 \frac{Deff t}{L^2}\right) \quad (2)$$

Where MR= unaccomplished moisture or moisture ratio. M_i = initial moisture content, M_t =moisture at time t , M_e = equilibrium moisture content. $Deff$ = effective moisture diffusivity (m^2/s), L = thickness of slab before drying, t = drying time [28].

$Deff$ is usually derived from experimental drying curves by plotting experimental values of $\ln MR$ versus time from equation (1) which gives a straight line graph with a slope of:

$$\text{Slope} = \frac{\pi^2 Deff}{L^2} \quad (3)$$

Temperature dependence of diffusivity has been shown to follow an Arrhenius relationship according to works done by [29]. The relationship is represented as;

$$D_{eff} = D_o \exp\left(\frac{E_a}{RT}\right) \quad (4)$$

Where, D_o = the pre-exponential factor of the Arrhenius equation (m^2/s), E_a =activation energy (kJ/mol K) and T = drying temperature ($^{\circ}C$). The activation energy is calculated by plotting $\ln Deff$ versus the reciprocal of the absolute temperature. Using equation (1) and (3), Despite the vast

research on the drying of pumpkin fruit slices, there is scarce literature on the drying behaviour of pumpkin (*Cucurbita pepo*) fruit slices in relation to sulphiting. This study aims at investigating the influence of sulphite on the internal moisture transfer mechanism during drying of pumpkin fruit slices, the drying time of fresh slices and reconstitution capacity of the dried pumpkin slices.

2. Materials and Methods

2.1. Raw Material

About 5 kg of matured pumpkin fruits of average moisture content $94.4\% \pm 0.21$ and $3.8^0 \pm 0.1$ Brix were harvested the same time from a private garden. These two parameters were used as indices for uniform maturity of the pumpkin fruits. The fruits were washed using tap water to remove soil and other debris. They were then kept in an incubated refrigerator at 4°C for temperature to equilibrate before use for the processing operations. Sodium metabisulphite (250 g) of laboratory grade was obtained from a chemical shop in Makurdi Benue state of Nigeria.

2.2. Sample Preparation

Pumpkin fruits were peeled using a stainless steel knife. The pulps of the fruits with seeds were removed and the thick mesocarp sliced in to pieces of 4 mm and 5 mm thickness with the aid of a digital vernier calliper. The choice of thinness was to simulate market samples based on the information obtained from the local producers. The sliced fruits were weighed in five portions of 50 g for each of the two sizes. Sodium metabisulphite solutions of 1.0, 1.5, 2.0 and 2.5% were prepared and placed in 500 ml beakers. To each concentration of salt, 50 g of 4 mm and 5 mm pumpkin fruit slices each were soaked in the prepared concentrations of sodium metabisulphate for 5 minutes to obtain four samples for the 4 mm size namely; P1, 4; P1.5, 4; P2, 4; P2.5, 4 and a non sulphited sample P0, 4 which was the control. Same thing was done for the 5 mm size of slices to obtain samples P1, 5; P1.5, 5; P2, 5; P2.5, 5 and a non sulphited sample P0, 5 was used as the control. The sample slices were removed, allowed to drain and arranged on non-perforated trays in a single uniformly thin layer within a hot air tray dryer so that hot air passed around one side of the slices. This was done for the controls as well which were soaked in water containing no salt as described by [30, 31]. Drying was carried out at 60, 65, 70, 75, and 80°C at a drying air velocity of 1.53 m/s, during which weights of the samples were taken at the intervals of 30 minutes until constant weight of sample was achieved. Samples were subjected to descriptive sensory evaluation to obtain sample P1, 4 dried at 75°C as the best among 4 mm thick samples and P2, 5 dried at 75°C as the best among 5 mm thick samples. Subsequent analysis were determined on these samples and their non sulphited counterparts (controls)

2.3. Drying Procedure

Moisture content (%) of samples was determined by material balance using data on weights generated during the drying process of the samples. Drying curves were drawn by plotting values of moisture ratio versus drying time as presented in Figure 1. Equilibrium moisture content (M_e) values were predicted graphically by plotting rate of drying (dx/dt) values of each sample against moisture content at a given temperature. M_e values were considered as the point where the Y-axis cuts X-axis. The ratio of the removed water to removable water in the sample known as moisture ratio (MR) was determined from equation. (1). Fick's unsteady state second law of diffusion equation (2) was used due to the long drying time to model all data as according to the method described by [24]. The effective moisture diffusivity (D_{eff}) was calculated from the slope of the curve plotted from linear values of moisture ratio ($\ln MR$) versus drying time (t) as given in equation (3) and the results were presented in Table 1. Activation energy (E_a) for each sample was determined using the Arrhenius temperature dependence of D_{eff} as given in equation (4). A plot of $\ln D_{eff}$ versus inverse of absolute temperature gave a slope from which (E_a) was calculated as shown in Table 2. Drying time was estimated using Fick's diffusion model for an infinite slab as given in equation (1). Equilibrium moisture content (M_e) values were predicted graphically by plotting rate of drying (dx/dt) values of each sample against moisture content at a given temperature. M_e values were considered as the point where the Y-axis cuts X axis. The result of drying time is presented in Table 3

2.4. Rehydration Capacity

Rehydration capacity of sulphited and dried pumpkin fruit slices was carried out with modifications according to the method described by [17]. In this method, 100 ml distilled water was added to 250 ml beaker and placed in a water bath at different rehydration temperatures of 30, 60 and 80°C at a time. To each beaker at its given temperature, 10 g of each sample P0, 4, P1, 4, P0, 5 and P2, 5 were rehydrated at a resident time of 15 minutes after which the slices were drained and reweighed. Rehydration capacity was calculated as:

$$RC = \frac{\text{Weight of soaked sample}}{\text{Weight of dry sample}}$$

The result of rehydration capacity is presented in Table 4.

3. Results and Discussion

3.1. Drying Kinetics

The drying curves of pumpkin fruit slices as shown in (Figures 1 and 2).

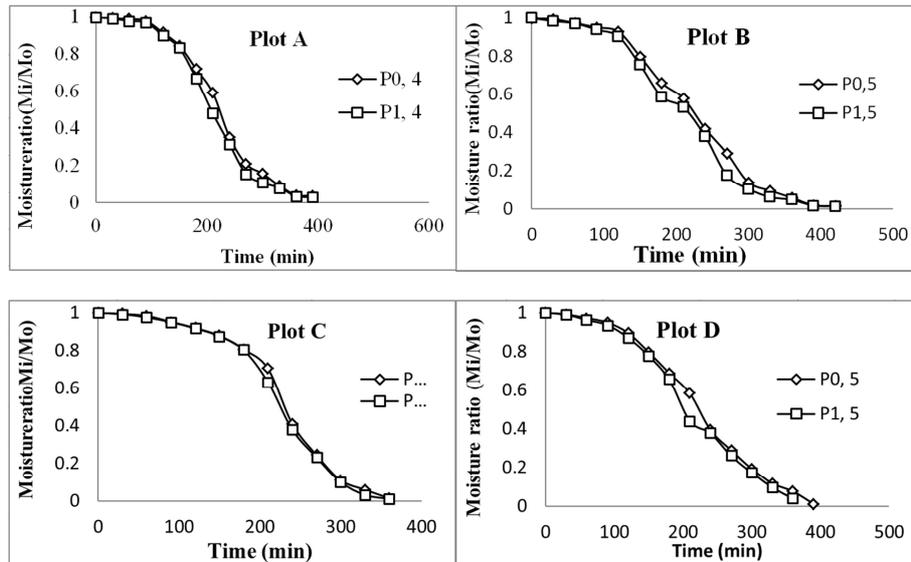


Figure 1. Effect of Sulphite on the Drying Behaviour of Pumpkin Fruit Slices of Thickness 4 mm (Plots A and C) and 5 mm (Plot B and D) at 60 and 65°C.

This is in agreement with the findings of [17] that dried strawberry pre-treated with alkaline ethyleolate solution in a laboratory dryer at temperatures of 50, 55 and 65°C. This could be attributed to the high initial moisture content (94.4%) of the pumpkin fruits. However, the falling rate dominated the drying process.

Drying predominantly followed a falling rate period. This was indicated by the presence of a short lived constant rate

period in the early stage of drying. It can be inferred that drying occurred by evaporation of water at the surface of the pumpkin slices as a result of free water that was available at a very high water activity. When the water activity reduced, the constant rate ceased and a falling rate period began. This implies that, moisture transfer during the falling rate period was predominantly caused by liquid diffusion or capillary flow.

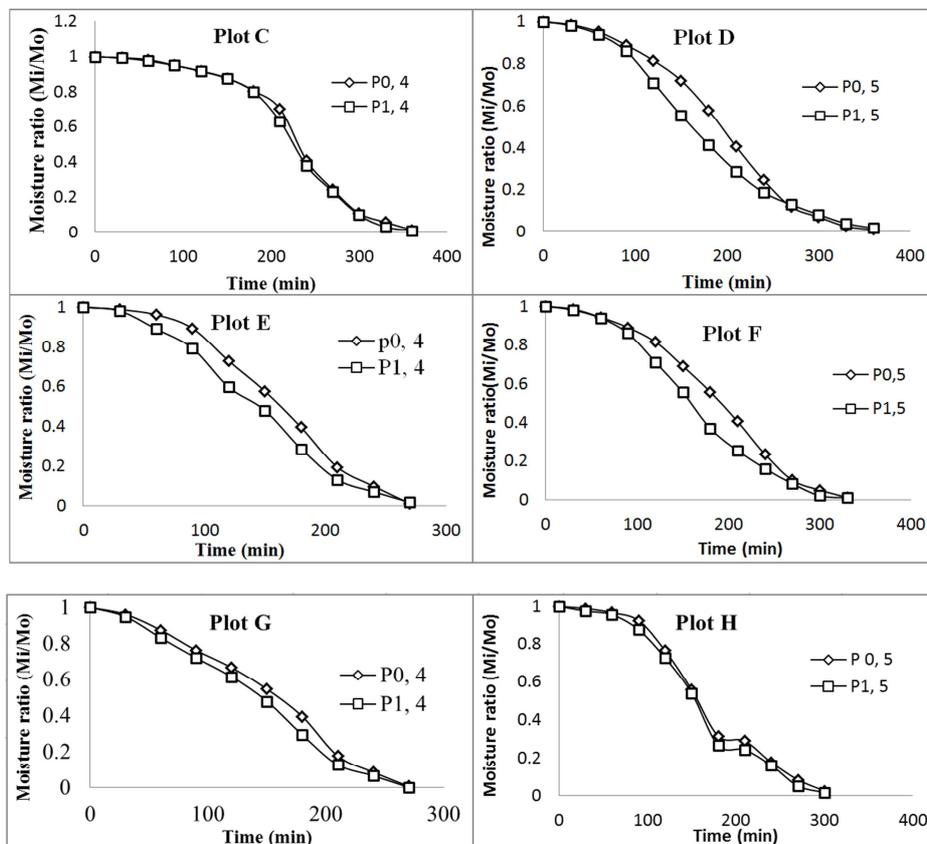


Figure 2. Effect of Sulphite on the Drying Behaviour of Pumpkin Fruit Slices of Thickness 4 mm (plots A, C and E) and 5 mm (plots B, D and F) at 70, 75 and 80°C.

A similar report was obtained by Tatsadjieu *et al*, who dried a high moisture (60.08%) food known as *kindumu* which is a fermented milk-based sorghum flour product [32]. Longer drying times were observed for the controls compared to the sulphited and the 5 mm thick compared to the 4 mm samples. This could be as a result of the influence of sulphite on the cellular membrane of the pumpkin fruit slices. Karim worked on the effect of pre-treatment on quality attributes of air dehydrated pineapple slices and reported a similar observation [33]. The shorter drying times in the 4 mm thick samples was as a result of increase in surface area due to size reduction [34].

3.2. Diffusivity (D_{eff})

The result of D_{eff} as presented in Table 1 revealed that diffusivity increased with increase in temperature for all the samples. This was in agreement with Roberts *et al* who dried three varieties of grape seeds and observed that effective diffusivity increased with increasing temperature [28]. The diffusivity values for the 4 mm slices ranged from 6.235×10^{-11} to $1.778 \times 10^{-10} \text{ m}^2/\text{s}$ while that of 5 mm slices ranged from 8.041×10^{-11} to $2.189 \times 10^{-10} \text{ m}^2/\text{s}$.

Table 1. Effect of Sulphiting on Moisture Diffusivity of 4 mm 5 mm Thick Pumpkin Fruit Slices.

Diffusivity ($\text{m}^2/\text{s} \times 10^{11}$)						
Samples		Temperatures ($^{\circ}\text{C}$)				
Sulphite (%)	Thickness (mm)	60	65	70	75	80
0.0	4	6.235	6.928	7.505	10.643	11.205
1.0	4	6.928	7.505	8.256	11.462	12.808
0.0	5	9.046	9.742	14.070	14.071	16.890
2.0	5	10.550	15.350	15.421	20.890	21.330

The determined values of D_{eff} for pumpkin fruit slices were in agreement with earlier reports that described the D_{eff} range for food materials to be from 8.0×10^{-12} to 6.46×10^{-9} and 10^{-11} to 10^{-9} [35, 17]. The control samples were observed to exhibit lower D_{eff} values than the sulphited samples. This could have been as a result of sulphite pre-treatment which might have affected the permeability of cellular membrane of the slices, creating a faster drying rate at the drying temperatures. This might have resulted in reduction of tentional and firmness of the slices thereby, causing an

increase in D_{eff} when compared to the control samples. This was also observed in papaya as well as sulphited and oven dried pineapple slices [36, 37].

3.3. Activation Energy

The activation energy values were estimated from the slopes of straight lines from plotting $\ln(D_{eff})$ versus $1/T$. The values of E_a for each sample thickness are shown in Table 2.

Table 2. Effect of Sulphiting on Arrhenius Parameters of Pumpkin Fruit Slices.

Regression parameter	Sulphite	Pumpkin Slices			
		0.0	1.0	0.0	2.0
	Thickness (mm)	4	4	5	5
n		5	5	5	5
r^2		0.947	0.956	0.974	0.945
Intercept		13.07	13.50	13.49	13.25
K_0		1.116	1.130	1.116	1.206
Slope		-3758	-3872	-3780	-3583
E_a (KJ/mol)		31.342	32.293	31.525	29.882

n= number of treatments, r^2 = coefficient of determination, k_0 = reaction rate constant, E_a = activation energy.

It was observed that the activation energy values increased with increasing slice thickness, showing the sensitivity of D_{eff} values to the slice thickness, i.e. the higher the activation energy, the higher the effect of sample thickness on D_{eff} values. E_a values for 4 mm slices were 31.342 and 32.293 kJ/mol for P0, 4 and P1, 4 respectively, while that of 5 mm slices were 31.525 and 29.882 kJ/mol for P0, 5 and P2, 5 respectively. These ranges fall within the general range of (15–40 kJ/mol) for food materials as reported by [38, 41].

3.4. Estimation of Drying Time

Drying time prediction based on Ficks second law of diffusion Table 3 showed that drying process was faster for the sulphited samples compared to their controls.

Table 3. Effect of Sulphiting and Slice Thickness on Drying Time of Pumpkin Fruit Slices.

Sample Drying Air Conditions Drying Time						
Code	Sulphiting (%)	Thickness (mm)	Temp. ($^{\circ}\text{C}$)	Velocity (m/s)	RH (%)	(hr)
P0, 4	0	4	75	1.52	32	16.5
P1, 4	1	4	75	1.52	32	15.2

Sample Drying Air Conditions Drying Time						
Code	Sulphiting (%)	Thickness (mm)	Temp. (°C)	Velocity (m/s)	RH (%)	(hr)
P0, 5	0	5	75	1.52	32	17.8
P2, 5	2	5	75	.52	32	16.9

P0, 4=Non sulphited pumpkin fruit slices of 4 mm, P1, 4=Pumpkin fruit slices of 4 mm thickness, 1% sulphited, P0, 5=Non sulphited pumpkin fruit slices of 5 mm thickness, P2, 5=Pumpkin fruit slices of 5 mm thickness, 2% sulphited.

The 5 mm thick sample took a longer time (16.9 h) to dry compared to the 4 mm sample (15.2 h). This implies that sample thickness had noticeable effect probably for the fact that drying of the sample took place in the falling rate period indicating that the mechanism of diffusion is governed by internal diffusion where the distant travelled by the diffusing vapour (thickness) becomes a controlling factor. The estimated drying time was higher than the experimental drying time for each of the sample thicknesses. This could be as a result of the fact that Ficks law of diffusion is based on assumptions which in practice do not hold, for instance, the model assumes negligible external resistance however, an air velocity of greater than 1 m/s has significant influence on drying [39].

3.5. Rehydration Capacity

Result of rehydration as presented in Table 4, showed that, the control samples rehydrated better than the sulphited samples at all the temperatures tested (30, 60 and 80°C). The 4 mm thick samples rehydrated better than the 5 mm thick samples. The 4 mm samples rehydrated best at 60°C followed by 30°C and least at 80°C. While the 5 mm samples rehydrated best at 30°C followed 60°C and the least rehydration ratio was observed at 80°C. It then implies that, the 4 mm samples can be rehydrated at 60°C while the 5 mm at 30°C. The 5 mm sample took a longer time to dry to its equilibrium moisture content and demonstrated lower rehydration ratios. Literature report revealed that the rehydration ratios of apple, potato, carrot, banana, pepper, garlic and mushroom range between 1 and 4, depending on the absorbed water temperature [40].

Table 4. Effect of Sulphiting and Slice Thickness on Rehydration Capacity of Dried Pumpkin Fruit Slices.

Sample	Temperature (°C)		
	30	60	80
P1, 4	3.72 ^{cd} ±0.03	3.98 ^a ±0.00	3.50 ^b ±0.03
P0, 4	3.94 ^a ±0.03	3.97 ^b ±0.01	3.78 ^a ±0.01
P2, 5	3.44 ^c ±0.02	3.14 ^c ±0.01	3.36 ^d ±0.02
P0, 5	3.62 ^d ±0.02	3.20 ^d ±0.01	3.28 ^c ±0.03
LSD	0.03	0.01	0.03

Values are means of triplicate determinations. Values with the same superscripts are not significantly different at $P < 0.05$. P0, 4=Non sulphited pumpkin fruit slices of 4 mm, P1, 4=Pumpkin fruit slices of 4 mm thickness, 1% sulphited, P0, 5=Non sulphited pumpkin fruit slices of 5 mm thickness, P2, 5=Pumpkin fruit slices of 5 mm thickness, 2% sulphited.

In this study, sulphited and dry pumpkin fruit slices had a rehydration ratio range of 3.14-3.98 at absorbed water temperatures of 30, 60 and 80°C which is within the reported range. It can therefore be inferred that the rehydration ratio of sulphited and dried pumpkin fruit slices was influenced by

rehydration temperature, pretreatment and sample thickness. This is in agreement with the report of Dhalsamant *et al* where rehydration ability of solar dried potato was significantly influenced by pretreatment followed by rehydration temperature and sample diameter [41].

4. Conclusion

Influence of sulphite on the drying kinetics of pumpkin revealed that, drying of sulphited pumpkin fruit slices occurred mainly in the falling rate period which is typical of the drying behaviour of food products. Diffusivity increased with increase in sulphite level and drying temperature for the 4 mm and 5 mm thick slices. Slice thickness also influenced the activation energy of pumpkin fruits slices. This implies that the internal transfer of moisture in the sulphited pumpkin slices was controlled by moisture diffusion and this is important in food packaging and storage where transfer of moisture, vapours (gases) and flavour compounds may adversely influence food quality. Sulphite enhanced reduced drying time of fresh pumpkin fruit slices as well as slice thickness. It can be inferred that drying time reductions in the drying process of fresh pumpkin fruit slices will definitely amount to significant energy savings since hot air drying processes do have intensive energy requirement. Based on the result of the reconstitution of dried pumpkin fruit slices, it can be deduced that, the non sulphited slices can be rehydrated better at room temperature (30°C) whereas the sulphited sample slices reconstitute better at temperatures higher than room temperature.

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