

# Environmental Health Status of Some Aquatic Ecosystems in Badagry Division, Lagos, Nigeria

Mekuleyi Gabriel Olarinde<sup>1,\*</sup>, Anetekhai Martins Agenuma<sup>1</sup>, Aderinola Oluwatoyin Joseph<sup>2</sup>, Adu Abosedé<sup>3</sup>

<sup>1</sup>Department of Fisheries, Faculty of Science, Lagos State University, Lagos, Nigeria

<sup>2</sup>Department of Zoology and Environmental Biology, Faculty of Science, Lagos State University, Lagos, Nigeria

<sup>3</sup>Department of Botany, Lagos State University, Ojo, Nigeria

## Email address:

gabrielmekuleyi@gmail.com (M. G. Olarinde), anetekhaimartins@gmail.com (A. M. Agenuma),

oluwatoyin.aderinola@lasu.edu.ng (A. O. Joseph), adu\_bose@yahoo.co.uk (A. Abosedé)

\*Corresponding author

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**Abstract:** The status of Agboju, Ajegunle, Topo and Gbaji water bodies in Badagry Division of Lagos State, Nigeria was evaluated between April and November 2018 in order to determine the suitability of the ecosystem to biota and safety of the aquatic resources for human consumption. Some environmental variables such as alkalinity, temperature, biochemical oxygen demand (BOD), conductivity, total dissolved solid (TDS), carbon (iv) oxide (CO<sub>2</sub>), total hardness, chemical oxygen demand (COD), turbidity, phosphate, dissolved oxygen (DO), pH, salinity, sulphate and nitrate, and some heavy metals (Fe, Cu, Zn, Pb, Cd and Cr) in water, sediment, aquatic plant (*Eichhornia crassipes*) and fish (*Ethmalosa fimbriata*) were measured using standard procedures. The results showed that BOD, COD, total hardness and turbidity values were above the standard permissible limits. Cd in water column of Agboju and Ajegunle, and Pb recorded from Agboju, Ajegunle and Topo, exceeded standard permissible limit. Cu (0.56±0.54), Zn (5.45±0.89), Pb (0.54±0.47) and Cr (0.455±0.375) contents in *E. fimbriata* (from Ajegunle) was slightly above permissible limits while Cd and Fe contents in the fish across the four stations were above permissible limits. Contamination factor, Index of geochemical accumulation and enrichment factor showed moderate degree of contamination of cadmium and Fe in the sediment. This preliminary study could be concluded that the biota can still survive in the ecosystems despite high contamination of the water bodies with metals. However, human consumption of several of *E. fimbriata* and water could cause health hazard. Therefore, urgent effort should be made by all users of the ecosystem to decline in discharging untreated waste into these water bodies.

**Keywords:** Aquatic Ecosystem, Badagry Division, Nigeria, Biota, Contamination

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## 1. Introduction

Productivity and sustainability of water are usually known by the physicochemical properties of water quality. Due to indiscriminate release of wastes, physico-chemical characteristics of water do changes and this could cause hazard effect to flora and fauna of the aquatic ecosystem. Naturally, waters contain some impurities such as metals whose nature and amount vary with source. Some trace metals like copper, zinc, cobalt, iron, and manganese are essential nutrients to

plants at very low concentrations. However, high concentrations of these metals are potential toxins to many biological systems. Reports have shown that high level of pollution with heavy metals result from terrigenous activities, atmospheric precipitation, industrial discharge or domestic sewage and non-point source run-off [1]. The hazard of an environmental chemical is a function of its environmental persistence, toxicity, and bio-accumulative potential. Hence, heavy metals are considered hazardous due to its persistence, bioaccumulation, and toxicity features. In many developing

countries, improper management of domestic waste, burning of fossil fuels, industrial wastes, agricultural run-off and mining of metals have created a potential source of pollution in the aquatic environment of many countries including Nigeria [2].

Some recent research findings have shown that there is continual pollution of Nigerian waters as attested to by high accumulation of toxic metals in fishes caught in them [3-10]. As a result, it becomes imperative to carry out periodic assessment of aquatic ecosystems in Nigeria.

Therefore, this study examined the physicochemical properties and concentrations of some heavy metals in water column, sediment, *Eichhornia crassipes* and *Ethmalosa fimbriata* from four aquatic ecosystems (Agboju, Gbaji, Topo and Ajegunle) within Badagry Division of Lagos State, Nigeria with a view to determine the suitability and safety of the biota in the water bodies as well as their safety for consumption.

## 2. Materials and Methods

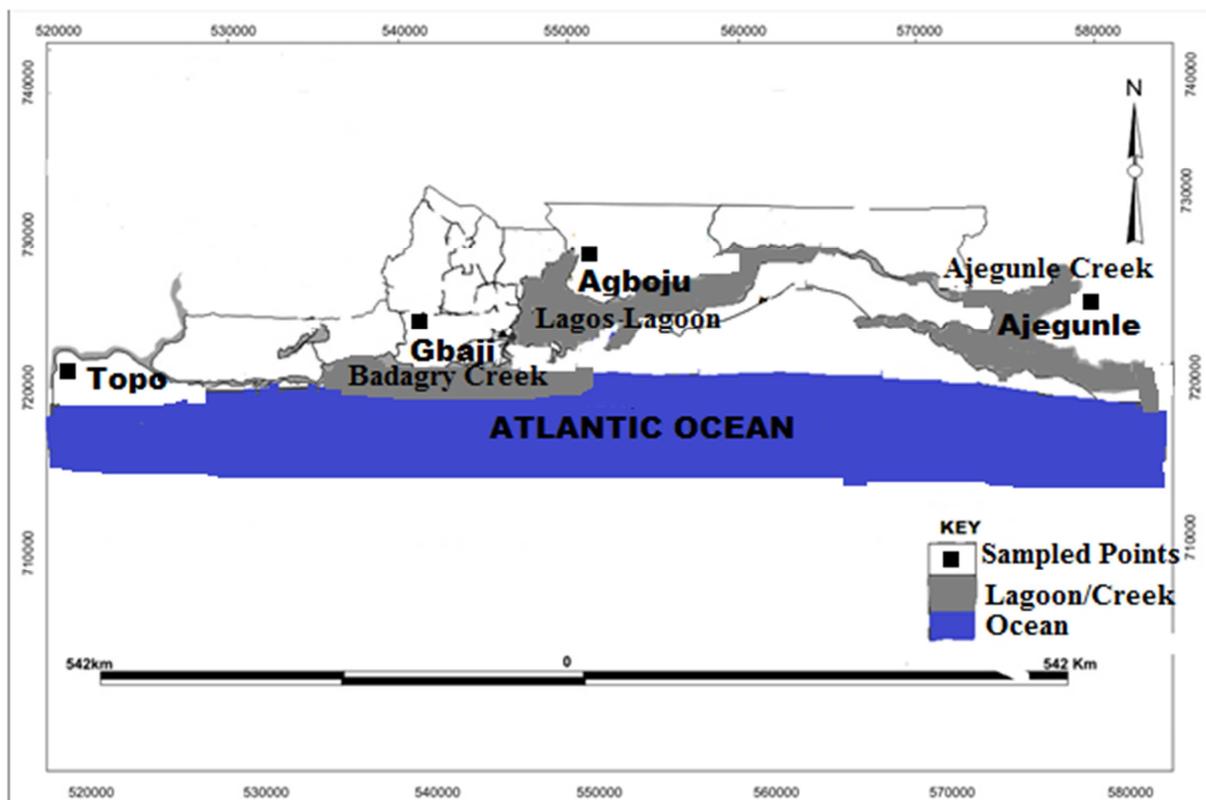
### 2.1. Study Area

Badagry Division lies between longitude  $2^{\circ}42'E$  and

$3^{\circ}22'E$  and between latitude  $6^{\circ}22'N$  and  $6^{\circ}42'N$ . It is bounded in the west by Republic of Benin and in south by the Atlantic Ocean. Also, it is bounded by Alimosho Local Government Area in the north but bounded in the east by Apapa Local Government Area. Badagry Division is surrounded by lagoons, rivers, streams, lakes and creeks.

Badagry Division in Lagos State, Nigeria consists of four local government areas namely Ojo, Amuwo-Odofin, Ajeromi- Ifelodun and Badagry. The study stations namely Gbaji and Topo (In Badagry Creek), Agboju (In Lagos Lagoon) and Ajegunle (In Ajegunle Creek) are as shown in Figure 1. Gbaji and Topo are located in Badagry Local Government Area. Agboju is located in Amuwo Odofin Local Government Area while Ajegunle is located in Ajeromi Ifelodun Local Government Area.

From record of national census in 2016, Amuwo Odofin LGA had a population of over 1,500,000, and shares its boundaries with Ajeromi Ifelodun which had population of 1,746,634 while Gbaji and Topo had a population of about 250,000 respectively.



**Figure 1.** Map Showing the location of Topo and Gbaji (In Badagry Creek), Agboju (In Lagos Lagoon) and Ajegunle (In Ajegunle Creek) across Badagry Division of Lagos State, Nigeria.

### 2.2. Collection and Pre-Treatment of Samples

Samples of water, 96 samples of sediment, 192 samples of *Eichhornia crassipes* and 192 samples of *Ethmalosa fimbriata* (Bonga fish) were collected from four sampling stations (Topo, Gbaji, Agboju and Ajegunle) in Badagry Division of Lagos State, Nigeria for six months. The study period comprises of 3

months of wet season (April – June, 2018) and dry season (September–November, 2018). Two (2) sampling points were chosen at each sampling station. Water was collected 20cm below the water surface from each sampling point twice every month by using acid-clean 2.5L polycarbonate bottles. From each water sample collected, 100 mL was measured and acidified with 0.5 mL of  $HNO_3$  to prevent microbial

degradation of the metals present in the water. Water sample was refrigerated at  $-4^{\circ}\text{C}$  before laboratory analysis.

Environmental variables (Physicochemical parameters) measured include alkalinity (mg/L), temperature ( $^{\circ}\text{C}$ ), biochemical oxygen demand (BOD, mg/L), electrical conductivity (EC,  $\mu\text{S}/\text{cm}$ ), total dissolved solids (TDS, mg/L), Carbon (iv) oxide ( $\text{CO}_2$ , mg/L), total hardness (mg/L), chemical oxygen demand (COD, mg/L), turbidity (NTU), phosphate (mg/L), dissolved oxygen (DO, mg/L), pH, salinity (ppt), sulphate (mg/L) and nitrate (mg/L). All the parameters were measured according to standard procedures [11-12]. All Samples of fresh *Ethmalosa fimbriata*, sediment and fresh

*Eichhornia crassipes* were collected and transported in polyethylene bags, previously cleaned and treated with 5%  $\text{HNO}_3$  and rinsed with distilled water [11]. The *E. fimbriata* and *E. crassipes* samples were washed with tap water and stored in a freezer at  $-4^{\circ}\text{C}$  for 96 hours.

### 2.3. Heavy Metal Analysis

All frozen samples were defrosted at  $25^{\circ}\text{C}$  attained in 5 hours. The heavy metals such as iron (Fe), copper (Cu), zinc (Zn), lead (Pb), cadmium (Cd) and chromium (Cr) concentrations in water samples were determined using Buck Scientific Atomic Absorption Spectrophotometer (VGP 210 model, USA). Sediment samples were dried with crucible for 12 min and thereafter allowed to cool and sieved with 0.5 mm sieve in preparation for digestion and heavy metal analysis. Metal content of the digested sediment samples was determined using Atomic Absorption Spectrophotometer (Buck scientific 210VGP model). 40 Samples of whole fresh *E. fimbriata* and *E. crassipes* were selected from each sampling station, weighed and oven-dried at  $105^{\circ}\text{C}$  for 28 hours till a constant weight was obtained. Dried samples were grounded with ceramic mortar and pestle, into powdery form and 3 g of each sample was digested according to APHA standard [11]. The concentration of the metals was calculated in mg/kg while the analytical procedure was checked using reference material (DORM 1, Institute of Environmental Chemistry, NRC Canada).

### 2.4. Enrichment Factor (EF)

Enrichment factor is used to estimate the anthropogenic impact on sediments [13]. The EF calculation reduces the metal variability associated with variations in sediment ratios. The EF method normalizes the measured heavy metal content with respect to a sample reference metal such as iron. In this study, sample reference metal used is iron (Fe) since Fe has a relatively high concentration in the earth while Gbaji is used as control station, as it contains element with low occurrence variability.

$$\text{EF} = (\text{C}_i/\text{C}_{ie}) \text{ s}/(\text{C}_i/\text{C}_{ie}) \text{ rs}$$

Where  $\text{C}_i$ =content of element i in the sample of interest or the selected reference sample.

$\text{C}_{ie}$ =content of immobile element in the sample or selected reference sample.

( $\text{C}_i/\text{C}_{ie}$ ) s=heavy metal to immobile element ratio in the

samples of interest or examined environment.

( $\text{C}_i/\text{C}_{ie}$ ) rs=heavy metal to immobile element ratio in the selected reference sample. Based on enrichment factor, 5 contamination categories are recognized namely  $\text{EF} < 2$ =minimal enrichment,  $2 \leq \text{EF} < 5$ =moderate enrichment,  $5 \leq \text{EF} < 20$ =significant enrichment,  $20 \leq \text{EF} < 40$ =very high enrichment, and  $\text{EF} > 40$ =extremely high enrichment.

### 2.5. Index of Geo-accumulation ( $I_{geo}$ )

Index of geo – accumulation is used in determining metal contamination in sediments, by comparing current concentration with pre- industrial levels [14].

$$I_{geo} = \text{Log } 2 (C_i/1.5 C_{ri})$$

Where  $\text{C}_i$ =measured concentration of the examined metal i in the sediment.

$\text{C}_{ri}$ =reference value of the metal i or geochemical background concentration of the metal i. Factor 1.5 helps to minimize possible variations in background values for a given metal. Background value for this study was considered from world average value in shale (mg/kg).

Geo-accumulation index has 7 categories namely:

Class 0= $I_{geo} \leq 0$  (practically uncontaminated), Class 1= $0 < I_{geo} < 1$  (uncontaminated to moderately contaminated), Class 2= $1 < I_{geo} < 2$  (moderately contaminated), Class 3= $2 < I_{geo} < 3$  (moderately to heavily contaminated), Class 4= $3 < I_{geo} < 4$  (heavily contaminated), Class 5= $4 < I_{geo} < 5$  (heavily to extremely contaminated), Class 6= $5 < I_{geo} > 6$  (extremely contaminated).

### 2.6. Contamination Factor

The Contamination Factor (CF) describes the level of contamination of a given metal in sediment over a period of time. CF is expressed as:  $\text{C}_m/\text{C}_b$ , where  $\text{C}_m$ =Concentration of a particular metal in sediment, and  $\text{C}_b$  is the background metal. The background for this study was taken from DPR [15].

$\text{CF} < 1$ =low degree of contamination,  $1 \leq \text{CF} < 3$ =moderate degree of contamination,  $3 \leq \text{CF} < 6$ =considerable degree of contamination, and  $\text{CF} \geq 6$ =very high degree of contamination.

### 2.7. Statistical Analyses

All Data for spatial and seasonal variations of the metals in the samples were tested by one-way Analysis of Variance (ANOVA) and results expressed as mean  $\pm$  standard deviation while Duncan Multiple Range Post-hoc test was used to separate the means. The level of significance was set at  $p < 0.05$ .

Also, Enrichment Factor, Contamination Factor (CF) and Index of geo – accumulation were analyzed using standard equations and formulars.

## 3. Results

### 3.1. Environmental Variables of Water Samples

The spatial variation of the environmental variables

(Physicochemical parameters) at each sampling site was presented in Table 1. There were no significant spatial variation ( $p>0.05$ ) in the values recorded for temperature, carbon (iv) oxide ( $\text{CO}_2$ ), phosphate, dissolved oxygen (DO), pH, salinity and sulphate across the four sampling stations. Gbaji had the highest value of alkalinity ( $38.44\pm 6.96$  mg/L) and total dissolved solid (TDS,  $135.89\pm 28.74$ mg/L) but lowest values in conductivity ( $220.27\pm 30.97$   $\mu\text{s/cm}$ ), total hardness ( $191.45\pm 10.89$ mg/L), Chemical oxygen demand (COD,  $166.21\pm 20.93$ mg/L) and turbidity ( $39.35\pm 6.26$  NTU). The highest Biochemical Oxygen Demand (BOD,  $217.50\pm 28.37$  mg/L), electrical conductivity ( $265.30\pm 45.41$   $\mu\text{s/cm}$ ), total hardness ( $238.25\pm 18.91$ mg/L), chemical oxygen demand (COD,  $180.46\pm 25.29$ ), turbidity ( $48.63\pm 7.08$  NTU) and nitrate ( $5.98\pm 0.84$ mg/L) were recorded in Ajegunle. However, Ajegunle had the lowest alkalinity ( $31.75\pm 4.21$ mg/L). On the contrary, the lowest TDS ( $116.61\pm 5.78$ mg/L) and lowest nitrate ( $4.30\pm 0.74$ mg/L) were obtained from Agboju, while the least value in BOD ( $205.46\pm 24.19$ mg/L) was recorded in Topo.

There were significant ( $p<0.05$ ) spatial variations between alkalinity values in Gbaji, Agboju and Topo in comparison with that of Ajegunle as shown in Table 1. Also, the spatial variations in the electrical conductivity, TDS and total hardness respectively among Agboju ( $254.37\pm 42.12$   $\mu\text{s/cm}$ ,  $116.61\pm 5.78$ mg/L and  $208.85\pm 14.76$ mg/L), Gbaji ( $220.27\pm 30.97$  $\mu\text{s/cm}$ ,  $135.89\pm 28.74$ mg/L and  $191.45\pm 10.89$ mg/L), Ajegunle ( $265.30\pm 45.41$  $\mu\text{s/cm}$ ,  $124.97\pm 4.38$ mg/L and  $238.25\pm 18.91$ mg/L) and Topo ( $236.60\pm 40.48$   $\mu\text{s/cm}$ ,  $130.98\pm 24.12$ mg/L and  $197.37\pm 5.53$ mg/L) were significant (Table 1). While only nitrate values recorded in Ajegunle was significantly ( $p<0.05$ ) high, the spatial variations of biochemical oxygen demand (BOD) in the sampling sites (except between Topo ( $205.46\pm 24.19$ mg/L) and Gbaji ( $205.88\pm 23.48$ )) was significant ( $p<0.05$ ). Similarly, there were no significant ( $p>0.05$ ) spatial variations in COD and Turbidity values between Gbaji ( $166.21\pm 20.93$ mg/L,  $39.35\pm 6.26$ NTU) and Topo ( $166.21\pm 20.93$ mg/L,  $40.68\pm 5.05$ NTU).

**Table 1.** Spatial Variation of Water Physico-chemical Parameters of Four Aquatic Ecosystems in Badagry Division, Lagos, Nigeria.

Parameters	Agboju	Gbaji	Ajegunle	Topo
Alkalinity (mg/L)	$37.37\pm 5.63^a$	$38.44\pm 6.96^a$	$31.75\pm 4.21^b$	$38.09\pm 6.49^a$
Temperature ( $^{\circ}\text{C}$ )	$28.03\pm 2.32^a$	$27.49\pm 1.99^a$	$28.63\pm 2.45^a$	$27.82\pm 2.11^a$
BOD (mg/L)	$212.31\pm 26.46^a$	$205.88\pm 23.48^b$	$217.50\pm 28.37^{ab}$	$205.46\pm 24.19^b$
Conductivity ( $\mu\text{s/cm}$ )	$254.37\pm 42.12^a$	$220.27\pm 30.97^b$	$265.30\pm 45.41^{ab}$	$236.60\pm 40.48^{ac}$
TDS (mg/L)	$116.61\pm 5.78^a$	$135.89\pm 28.74^b$	$124.97\pm 4.38^{ab}$	$130.98\pm 24.12^{ac}$
$\text{CO}_2$ (mg/L)	$8.73\pm 1.65^a$	$8.38\pm 2.11^a$	$9.77\pm 2.09^a$	$8.40\pm 2.17^a$
Total Hardness (mg/L)	$208.85\pm 14.76^a$	$191.45\pm 10.89^b$	$238.25\pm 18.91^{ab}$	$197.37\pm 5.53^{ac}$
COD (mg/L)	$175.90\pm 21.65^a$	$166.21\pm 20.93^b$	$180.46\pm 25.29^{ab}$	$167.50\pm 20.44^b$
Turbidity (NTU)	$45.43\pm 6.76^a$	$39.35\pm 6.26^b$	$48.63\pm 7.08^{ab}$	$40.68\pm 5.05^b$
Phosphate (mg/L)	$4.56\pm 0.84^a$	$4.34\pm 0.66^a$	$5.17\pm 1.03^a$	$4.41\pm 0.78^a$
DO (mg/L)	$6.20\pm 0.61^a$	$6.33\pm 0.88^a$	$5.77\pm 0.78^a$	$6.76\pm 0.99^a$
pH (mg/L)	$7.18\pm 0.54^a$	$7.32\pm 0.44^a$	$7.06\pm 0.53^a$	$7.43\pm 0.52^a$
Salinity (ppt)	$0.037\pm 0.032^a$	$0.081\pm 0.099^a$	$0.040\pm 0.040^a$	$0.068\pm 0.067^a$
Sulphate (mg/L)	$4.29\pm 0.88^a$	$4.26\pm 0.43^a$	$4.43\pm 0.59^a$	$4.11\pm 0.41^a$
Nitrate (mg/L)	$4.30\pm 0.74^a$	$4.98\pm 0.82^a$	$5.98\pm 0.84^{ab}$	$4.54\pm 0.89^a$

Mean  $\pm$  SD values with different superscript in the row are significantly ( $p<0.05$ ) different.

The respective wet and dry seasonal variations for alkalinity ( $38.59\pm 5.94$ mg/L,  $34.23\pm 5.81$ mg/L), temperature ( $26.41\pm 0.93^{\circ}\text{C}$ ,  $29.57\pm 1.75^{\circ}\text{C}$ ), BOD ( $187.73\pm 3.84$ mg/L,  $232.84\pm 11.42$ mg/L), conductivity ( $277.74\pm 30.76$  $\mu\text{s/cm}$ ,  $210.50\pm 12.44$  $\mu\text{s/cm}$ ) TDS ( $139.14\pm 19.43$ mg/L,  $115.09\pm 9.13$ mg/L),  $\text{CO}_2$  ( $7.11\pm 0.73$ mg/L,  $10.52\pm 1.09$ mg/L), total hardness ( $205.38\pm 27.52$ mg/L,  $212.56\pm 15.88$ mg/L), COD ( $156.03\pm 6.19$ mg/L,  $189.00\pm 18.36$ mg/L), turbidity ( $39.19\pm 5.80$ NTU,  $47.86\pm 5.34$ NTU), phosphate ( $4.77\pm 1.02$ mg/L,  $4.47\pm 0.65$ mg/L), DO ( $6.78\pm 0.72$ mg/L,  $5.75\pm 0.64$ mg/L), pH ( $7.56\pm 0.44$ ,  $6.93\pm 0.32$ ), salinity ( $0.093\pm 0.074$ ppt,  $0.021\pm 0.012$ ppt), sulphate ( $4.68\pm 0.45$ mg/L,  $3.87\pm 0.38$ mg/L) and nitrate ( $5.34\pm 1.04$ mg/L,  $4.56\pm 0.84$ mg/L) are as shown in Figure 2. The parameters (except  $\text{CO}_2$ , phosphate, DO, pH, salinity, sulphate and nitrate) showed significant ( $p<0.05$ ) seasonal variation across the sampling stations. While alkalinity, conductivity and TDS were higher in wet seasons, temperature, BOD, total hardness, COD and turbidity were significantly high during dry seasons.

### 3.2. Heavy Metals Concentration in Water Sample and *Eichhornia Crassipes*

There were no significant ( $p>0.05$ ) spatial and seasonal variation in heavy metals concentration in water column recorded in this study. The metals recorded in wet season and dry season respectively in the water samples are Fe ( $0.96\pm 0.28$ mg/L,  $1.47\pm 0.45$ mg/L), Cu ( $0.23\pm 0.14$ mg/L,  $0.59\pm 0.28$ mg/L), Zn ( $0.70\pm 0.48$ mg/L,  $0.89\pm 0.37$ mg/L), Pb ( $0.045\pm 0.033$ mg/L,  $0.118\pm 0.111$ mg/L), Cd ( $0.003\pm 0.003$ mg/L,  $0.007\pm 0.003$ mg/L) and Cr ( $0.005\pm 0.003$ mg/L,  $0.007\pm 0.007$ mg/L). The spatial heavy metals in water sample of Agboju, Gbaji, Ajegunle and Topo respectively are: Fe ( $1.13\pm 0.59$ mg/L,  $1.09\pm 0.28$ mg/L,  $1.56\pm 0.45$ mg/L,  $1.10\pm 0.25$ mg/L), Cu ( $0.29\pm 0.27$ mg/L,  $0.28\pm 0.17$ mg/L,  $0.60\pm 0.47$ mg/L,  $0.30\pm 0.28$ mg/L), Zn ( $0.81\pm 0.29$ mg/L,  $0.62\pm 0.34$ mg/L,  $1.29\pm 0.39$ mg/L,  $0.49\pm 0.27$ mg/L), Pb ( $0.074\pm 0.070$ mg/L,  $0.027\pm 0.019$ mg/L,  $0.154\pm 0.164$ mg/L,  $0.058\pm 0.043$ mg/L), Cd ( $0.004\pm 0.003$ mg/L,

0.003±0.002mg/L, 0.009±0.006mg/L, 0.003±0.001mg/L) and Cr (0.004±0.002mg/L, 0.004±0.003mg/L, 0.011±0.009mg/L, 0.005±0.003mg/L).

Only Fe and Zn content in the *E. crassipes* had spatial significant difference ( $p < 0.05$ ) across the sampling stations. The highest values of Fe (270.89±41.25mg/kg) and Zn (11.09±3.67mg/kg) were recorded in Ajegunle while the lowest Fe (108.19±9.82mg/kg) and Zn (8.92±5.99mg/kg) was obtained from Gbaji. However, there were no significant ( $p > 0.05$ ) differences between the Fe and Zn contents recorded in Gbaji and Topo. While the values of Pb (1.12±0.60mg/kg, 1.39±0.55mg/kg), Cd (0.59±0.28, 0.90±0.49) Cr (0.34±0.33, 0.90±0.52) and Cu (1.49±0.57, 1.27±0.32) during wet and dry season respectively, had no significant ( $p > 0.05$ ) variation, other metals (Fe and Zn) recorded in *E. crassipes* showed significant ( $p < 0.05$ ) seasonal variation across the sampling sites. The peak value of Fe (178.72±84.05mg/kg) and Zn (14.37±3.54mg/kg) in *E. crassipes* were obtained during dry season while the lower values of Fe (167.78±61.48mg/kg) and Zn (5.86±1.85mg/kg) in *E. crassipes* were recorded in wet season.

### 3.3. Heavy Metal Content in Sediment

Only Fe, Cu and Zn showed significant ( $p < 0.05$ ) spatial variation of heavy metals content in the sediment examined from the four sampling stations (Table 2). The peak Fe (1405.51±126.23mg/kg), Cu (11.96±2.62mg/kg) and Zn (19.15±4.66mg/kg) contents in the sediment was obtained

from Ajegunle. On the other hand, the lowest Fe (1056.50±100.09mg/kg) was recorded in Gbaji, while lowest Cu (7.74±1.73mg/kg) and Zn (12.84±1.77mg/kg) were obtained from Topo. Similarly, significant ( $p < 0.05$ ) seasonal variation were recorded only for Fe, Cu and Zn in this study. The lowest seasonal values of Fe (1143.75±169.26mg/kg), Cu (8.66±2.21mg/kg), and Zn (13.73±2.33mg/kg) in the sediment were recorded in wet season, while the higher seasonal values of Fe (1260.35±189.07mg/kg), Cu (10.67±3.15mg/kg) and Zn (17.89±4.84mg/kg) were obtained in dry season. The values of other metals in sediment during wet and dry season respectively are Pb (1.47±0.70mg/kg, 1.59±0.94mg/kg), Cd (1.16±0.36mg/kg, 1.13±0.58mg/kg) and Cr (1.14±0.34mg/kg, 1.25±0.47mg/kg).

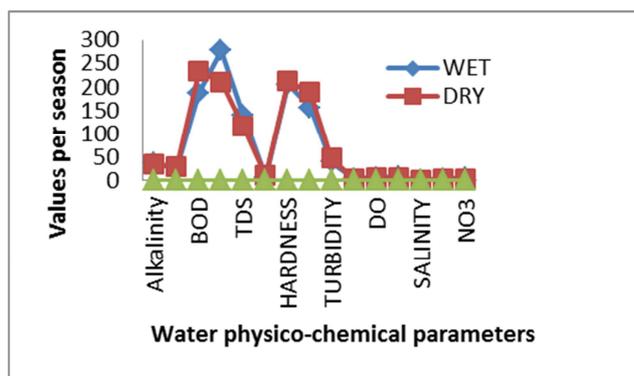


Figure 2. Seasonal Variation of Water Physico-chemical Parameters.

Table 2. Spatial Variation of Heavy Metals Concentration in Sediment from Four Aquatic Ecosystems in Badagry Division, Lagos, Nigeria.

Heavy Metals	Agboju	Gbaji	Ajegunle	Topo
Fe (mg/kg)	1347.39±116.38 <sup>a</sup>	1056.50±100.09 <sup>b</sup>	1405.51±126.23 <sup>ab</sup>	1063.07±94.62 <sup>ac</sup>
Cu (mg/kg)	10.83±3.23 <sup>ab</sup>	8.13±1.14 <sup>b</sup>	11.96±2.62 <sup>ab</sup>	7.74±1.73 <sup>b</sup>
Zn (mg/kg)	17.72±4.33 <sup>a</sup>	13.54±2.06 <sup>b</sup>	19.15±4.66 <sup>a</sup>	12.84±1.77 <sup>b</sup>
Pb (mg/kg)	1.50±0.44 <sup>a</sup>	1.22±0.31 <sup>a</sup>	2.34±1.25 <sup>a</sup>	1.07±0.16 <sup>a</sup>
Cd (mg/kg)	1.22±0.19 <sup>a</sup>	0.83±0.22 <sup>a</sup>	1.68±0.58 <sup>a</sup>	0.85±0.17 <sup>a</sup>
Cr (mg/kg)	1.33±0.35 <sup>a</sup>	0.92±0.18 <sup>a</sup>	1.65±0.30 <sup>a</sup>	0.91±0.22 <sup>a</sup>

Mean ± SD values with same superscript in the row are not significantly ( $p > 0.05$ ) different.

### 3.4. Enrichment Factor, Index of Geochemical

Accumulation and Contamination Factor of Sediment Sample

The values of enrichment factor (EF), index of geochemical accumulation (Igeo) and contamination factor (CF) of the sediment are presented in Figure 3, Tables 3 and 4 respectively. The computed EF values of heavy metals for Agboju are: Fe (1.000), Cu (1.045), Zn (1.026), Pb (0.964), Cd (1.153) and Cr (1.134). Similarly, Ajegunle sediment had EF values of Fe (1.000), Cu (1.106), Zn (1.063), Pb (1.442), Cd (1.521) and Cr (1.348) while EF obtained from Topo are Fe (1.000), Cu (0.945), Zn (0.942), Pb (0.872), Cd (1.018) and Cr (0.983). Only I<sub>geo</sub> of Cd (1.12373) at Ajegunle was under class 1 while other I<sub>geo</sub> of other metals from the four stations fall within Class 0. The values of CF in Cd obtained in sediment from the four sampling stations are: Agboju (1.525), Gbaji (1.038), Ajegunle (2.100) and Topo (1.062). However, the CF values of other metals (Fe, Zn, Cu, Pb and Cr) across the

stations were <1 (Table 4).

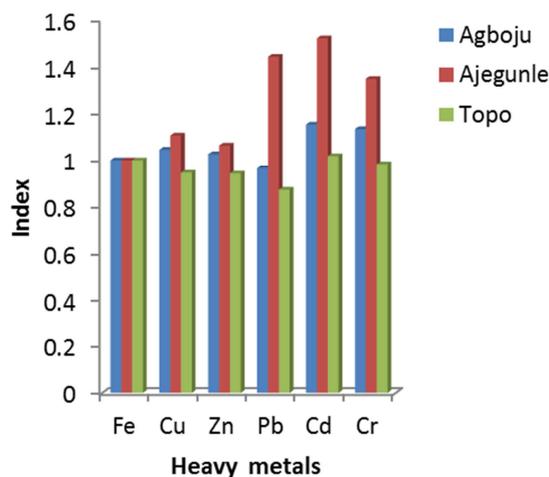


Figure 3. Enrichment Factor (EF) at Agboju, Ajegunle and Topo in Badagry Division, Lagos, Nigeria.

**Table 3.** Geo-Accumulation Index in Sediment from Four Aquatic Ecosystems in Badagry Division, Lagos, Nigeria.

Heavy Metals	Agboju ( $I_{geo}$ )	Gbaji ( $I_{geo}$ )	Ajgunle ( $I_{geo}$ )	Topo ( $I_{geo}$ )	World average value of metals (mg/kg)
Fe	0.00579	0.00454	0.00604	0.00457	46,700
Cu	0.19757	0.14831	0.21818	0.14119	45
Zn	0.03743	0.02860	0.04045	0.02712	95
Pb	0.01505	0.01224	0.02348	0.01074	20
Cd	0.81604	0.55518	1.12373	0.56856	0.3
Cr	0.00297	0.00205	0.00368	0.00203	90

**Table 4.** Contamination Factor (CF) in Sediment from Four Aquatic Ecosystems in Badagry Division, Lagos, Nigeria.

Metals	Background value	$C_f$ (Agboju)	$C_f$ (Gbaji)	$C_f$ (Ajgunle)	$C_f$ (Topo)
Fe	38000	0.035	0.028	0.037	0.028
Cu	36	0.301	0.226	0.332	0.215
Zn	140	0.127	0.097	0.137	0.092
Pb	85	0.018	0.014	0.028	0.013
Cd	0.8	1.525	1.038	2.100	1.062
Cr	100	0.013	0.009	0.017	0.009

### 3.5. Concentration of Heavy Metals in *Ethmalosa Fimbriata*

Only Zn and Fe showed spatial variation of heavy metal content in *Ethmalosa fimbriata* examined in this study. The value of Zn ( $5.45 \pm 0.89$  mg/kg) recorded from Ajgunle was significantly ( $p < 0.05$ ) higher than that from Agboju ( $3.88 \pm 0.91$  mg/kg), Gbaji ( $3.87 \pm 0.75$  mg/kg) and Topo ( $3.86 \pm 0.92$  mg/kg). Similarly, the value of Fe ( $11.63 \pm 3.20$  mg/kg) obtained in *E. fimbriata* from Ajgunle and Agboju ( $10.86 \pm 2.90$  mg/kg) were higher than that from Gbaji ( $7.46 \pm 3.01$  mg/kg) and Topo ( $7.39 \pm 3.70$  mg/kg). Other metals (in mg/kg) detected in *E. fimbriata* from Agboju, Gbaji, Ajgunle and Topo respectively are Cu ( $0.44 \pm 0.34$ ,  $0.32 \pm 0.19$ ,  $0.56 \pm 0.54$ ,  $0.06 \pm 0.04$ ), Pb ( $0.58 \pm 0.25$ ,  $0.20 \pm 0.11$ ,  $0.54 \pm 0.47$ ,  $0.03 \pm 0.02$ ), Cd ( $0.050 \pm 0.044$ ,  $0.031 \pm 0.029$ ,  $0.122 \pm 0.116$ ,  $0.081 \pm 0.054$ ) and Cr ( $0.070 \pm 0.057$ ,  $0.026 \pm 0.023$ ,  $0.455 \pm 0.375$ ,  $0.087 \pm 0.064$ ). The value of heavy metals recorded in *E. fimbriata* in dry and wet seasons respectively are Fe ( $11.70 \pm 2.64$  mg/kg,  $6.97 \pm 2.99$  mg/kg), Cu ( $0.52 \pm 0.42$  mg/kg,  $0.22 \pm 0.16$  mg/kg), Zn ( $4.72 \pm 1.03$  mg/kg,  $3.81 \pm 0.96$  mg/kg), Pb ( $0.53 \pm 0.49$  mg/kg,  $0.18 \pm 0.10$  mg/kg), Cd ( $0.105 \pm 0.089$  mg/kg,  $0.037 \pm 0.036$  mg/kg) and Cr ( $0.326 \pm 0.235$  mg/kg,  $0.132 \pm 0.066$  mg/kg). However, only Fe showed significant ( $p < 0.05$ ) seasonal variation with the higher value ( $11.70 \pm 2.64$  mg/kg) in the fish obtained during dry season.

## 4. Discussion

### 4.1. Environmental Variables and Heavy Metals in Water Samples

Only eight parameters (alkalinity, electrical conductivity, TDS, total hardness, BOD, COD, turbidity and nitrate) showed significant spatial variation across the four sampling stations. These findings could suggest that the four aquatic ecosystems exhibit similar physico-chemical properties. The suitability of parameters such as DO, pH, temperature,  $CO_2$ , and conductivity in this study could be the basis for the successful fisheries in these water bodies. All the

physicochemical parameters (except BOD, COD, total hardness and turbidity) are below the recommended maximum permissible limits [16-18]. All the values recorded for BOD, total hardness, COD and turbidity exceeded 6mg/L, 100mg/L, 10-40mg/L and 5NTU respectively of permissible limit in water. High BOD, COD, turbidity and hardness is an indication that the water from the four stations has poor quality. The parameters (except alkalinity, hardness, BOD, conductivity and COD) in this study were higher than that reported in Badagry creek [19]. It was also higher than that reported in Cross River Estuary, in Ogbe creek, in Tomaro creek, Tidal creek and from Asejire Lake respectively [10, 20-23]. On the other hand, the values of phosphate, hardness, salinity and TDS in this study were lower than that recorded in Majidun River in Lagos, Nigeria [24]. Similarly, values of all physico-chemical parameters in this study were lower than that reported in Ologe Lagoon, Ojo creek, and Lena River respectively [25-27]. Apart from  $CO_2$ , phosphate, DO, pH, salinity, sulphate and nitrate, other parameters showed significant seasonal variation across the sampling stations. Similar seasonal variation has been reported in Ologe Lagoon [25, 28]. However, no seasonal variation was reported in DO, pH and salinity in water from Nun and Brass River in Bayelsa State, Nigeria [29].

In this study, alkalinity, conductivity and TDS were higher in wet seasons, while temperature, BOD, total hardness, COD and turbidity were significantly high during dry seasons. In contrast to our findings, high pH, salinity, free carbon (iv) oxide, sulphate and total hardness has been recorded in wet season from Badagry creek [30].

The spatial and seasonal values of heavy metals concentration of water recorded in this study were not significant. This could imply that the water bodies have similar levels of metal contamination. Only values of Cu, Fe, Cr and Zn were below standard maximum permissible limit. This finding differs from other studies that reported higher concentration of the metals in surface water during the dry season [31-32]. The values of Cd in water column of Agboju and Ajgunle, and Pb recorded from Agboju, Ajgunle and Topo, exceeded the recommended permissible limit of Cd

(0.04mg/L) and Pb (0.003mg/L)[16, 33-34]. However, the Cd recorded in water for this study was lower than that reported in River Benue in Nigeria [35]. In contrast to the present findings, Cd concentration in surface waters that are within the acceptable limit for drinking water quality has been reported [36-37]. Also, high Cd concentration has been reported by some authors during the rainy season and during dry season respectively [37-38].

Studies have shown that cadmium (Cd) can easily contaminate essential amino acids and accumulates in the proximal tubular cells in higher concentrations and thus causes fragile bones and morphopathological changes in kidney and lung. High Cd could also cause liver dysfunction and neurotoxin, as well as reduced birth weight and premature delivery to pregnant women [39]. The values of Pb in the water column from Ajegunle, Topo and Agboju which exceeded the standard limit corroborated with other studies that reported Pb above permissible limits such as in Badagry creek and River Benue [19, 40]. High concentration of Pb has been reported to be responsible for chronic neurological disorders in infants and children [41]. In this study, the presence of Fe, Cu, Pb Cr, Cd and Zn in the water column is an indication that the aquatic resources in the four aquatic stations are exposed to pollution.

#### 4.2. Heavy Metals in *Ethmalosa Fimbriata*

Spatial and seasonal variation in *Ethmalosa fimbriata* was recorded only for Zn and Fe contents in this study. The values of Cu, Zn and Cr contents in *E. fimbriata* (from Ajegunle) exceeded the permissible limits of 0.4mg/kg, 5.0mg/kg and 0.10mg/kg for Cu, Zn and Cr respectively [42]. Similarly, the values of Pb in the fish from Ajegunle and Agboju were above the permissible limit of 0.29mg/kg. However, the Cd and Fe contents in the fish across the four stations were above permissible limits of 0.002mg/kg and 3mg/kg for Cd and Fe respectively [42]. Similar concentrations of some heavy metals that exceeded the limit in *Chrysichthys nigrodigitatus* from Ibaka and Ifiayong fishing sites in Akwa Ibom State has been reported [4]. The values of Pb, Cr, Zn and Cd in this study was higher than that recorded in *E. fimbriata* from Cross River Estuary [10]. But the values were lower than that reported in *Pseudotolithus senegalensis* and *E. fimbriata* from Carter Bridge River and Makoko River in Lagos, Nigeria [43]. The values of metals recorded in fish in this study were also lower than the values reported in *Cynothrissa mento* from Ologe Lagoon in Lagos Nigeria, in *E. fimbriata* and *Tilapia guineensis* from Imo River, in *Parachana obscura* from Epe Lagoon, in *Oreochromis mossambicus* from Malir river, Karachi and in *Oreochromis niloticus* from Asejire creek respectively [6-9, 23]. The value of Cu in fish for this study was lower than that reported in *C. nigrodigitatus* [4] from Ibaka and Ifiayong fishing sites in Akwa Ibom State.

Contamination of aquatic habitat with heavy metals like cadmium, lead, copper and chromium has significant health hazard to human beings through food chain [44]. Studies have shown that concentration of cadmium beyond tolerable limits result in reproductive, behavioral, cardiovascular,

neurological, hematological, and renal effects, damage of liver, kidney and testicles [45].

Various research studies indicated adverse effects of chromium in fish and such effects include decrease in hemoglobin and total erythrocytes count, thrombocytopenia, decline in proteins, lipids and the contents of glycogen [46].

The clogging action caused by excess iron precipitate hinders the respiration of fishes. The toxicity of iron on human cells has led to iron mediated tissue damage involving cellular reducing and oxidizing mechanism and their toxicity towards intracellular organelles such as lysosomes and mitochondria [47].

Lead toxicity in human causes changes in various biological processes such as cell adhesion, release of neurotransmitters, intra- and inter-cellular signaling, protein folding, maturation, apoptosis, ionic transportation, and enzyme regulation.

Zinc contamination has effect on the hepatic distribution of other trace metals in fish. Its deficiency has been associated with hypogonadism with impaired reproductive capacity, growth retardation, dermatitis, anorexia, and poor wound healing.

Although copper is an essential substance to human life, however, in high concentrations, it can cause intestinal irritation, anemia, liver and kidney damage [48].

#### 4.3. Heavy Metals in *Eichhornia Crassipes*

Only Fe and Zn contents in *E. crassipes* had significant spatial and seasonal variation in this study. The high concentration of iron and zinc in *E. crassipes* could be attributed to the fact that Fe and Zn are abundant in the water bodies. The observed non significant concentrations of Cd, Pb, Cr and Cu in the *E. crassipes* across the sampling stations could be due to efficient mechanism of elimination by this plant. Studies have shown that *E. crassipes* have the efficacy to decontaminate soil and water due to its high affinity to accumulate heavy metals in their root systems [49-50]. The level of all the metals detected in *E. crassipes* across the sampling stations were below WHO recommended limit of metals concentration in plant [51].

#### 4.4. Heavy Metals Concentration in Sediments

Only Fe, Cu and Zn show significant spatial and seasonal variation of heavy metals content in the sediment examined from the four sampling stations. The concentrations of these metals were higher in dry season and this could be due to the lower water flow during dry season that may facilitate accumulation of the heavy metals in sediment. Similar seasonal variation was reported in Bonny River, Nigeria [52]. Recommended maximum limits of metals in sediments are: Fe (30mg/kg), Pb (40mg/kg), Zn (123 mg/kg), Cu (25 mg/kg), Cr (20mg/kg) and Cd (0.5mg/kg) [33]. All the heavy metals (except Fe and Cd) in the sediment were below the recommended limit. High concentration of lead and iron could be from automobile emissions, metallic pipes, batteries of pumping equipment and as well as drainage of lead

compounds from adjacent crop farms treated with agro-chemicals. However, the present of Fe and Cd in the sediments above standard limit indicated that the sediment is polluted with Fe and Cd. All the metals in the sediment were lower than that reported in Bonny River, Calabar River and Euphrate River, and in Awoye Estuary in Ilaje Area of Ondo State [52-53]. It is clear that sediments act as the main pool of metals in the aquatic environment. Thus, their quality can indicate the status of water pollution.

#### 4.5. Evaluation of Contamination in Sediment Using Various Pollution Indices

The values of contamination factor (CF) in Cd obtained in sediment from Agboju, Gbaji, Ajegunle and Topo indicated a moderate degree of contamination of sediment with cadmium. However, the CF values of other metals (Fe, Zn, Cu, Pb and Cr) across the stations indicated low degree of contamination. This findings was different from the report on Abattoir sediment from PortHarcourt in which the sediment was not contaminated by Fe, Cu, Pb, Cr and Cd [54].

All the Igeo values (except that of Cd) were within zero category and this indicated that the sediment in this study were practically uncontaminated to moderately contaminated with the metals. On the other hand, Igeo values of Cd in Ajegunle falls within class 1 and this imply that the sediment were moderately contaminated with lead. All the values for enrichment factors in this study indicated minimal enrichment. In contrast, EF values of Pb, Fe and Cd that indicated high pollution of sediment was reported from Soku Oil field in Niger Delta [55]. Also, poor enrichment of metals in the sediment of Bonny River has been documented [53].

## 5. Conclusion

The present study has shown that BOD, COD, total hardness and turbidity, Cd and Fe contents in the fish across the four stations were above standard permissible limits. This finding indicated that the water bodies have poor quality, with the poorest quality being recorded in Ajegunle. Although, the contamination factor (CF), geoaccumulation index (Igeo) and enrichment index (EF) in sediment across the water bodies indicated low degree of pollution of the water bodies, however, human consumption of several of the fish species examined and the water, especially from Ajegunle creek could lead to Cd and Pb biomagnification and eventually could cause Cd and Pb related diseases. Therefore, urgent effort should be made by all users of the ecosystem to decline in discharging untreated waste into these water bodies.

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